

GCSE Chemistry B (Twenty First Century Science)

J258/03 Breadth in chemistry (Higher Tier)

Question Set 23

- 1** Mia has three metals, **A**, **B** and **C**, that she reacts with water.
- This is what she sees:
- Metal **A** Fizzes and reacts quickly.
- Metal **B** A few bubbles appear after some time.
- Metal **C** Slow fizzing.
- (a) (i) Which metal forms positive ions most easily?
Give **one** reason for your answer. [1]
- (ii) Metal **A** reacts with water to form **A²⁺** ions.
Write the **half** equation for this reaction. [1]
- (iii) Explain whether metal **A** is being reduced or oxidised. [1]
- (b) Metal **C** goes brown when put in copper sulfate solution, CuSO_4 .
Metal **C** forms a compound that contains **C²⁺** ions.
Write a balanced symbol equation for this reaction. Include state symbols. [2]
- (c) Metals conduct electricity.
Describe how the bonding in metals explains this. [2]

Total Marks for Question Set 23: 7

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